

Review of developing of CO₂ sequestration technology in coal seams

LETÍCIA TEIXEIRA PALLA BRAGA

The Strata Mechanics Research Institute of the Polish Academy of Sciences, Reymonta 27, 30-059 Krakow, Poland
braga@img-pan.krakow.pl

Abstract

Considering the global race against the increase of CO₂ emissions in the atmosphere, mitigation has been strongly advised and several technologies have been tested. In this context, research on CO₂ sequestration in coal seams using ECBM technology is emerging.

This work is a review that aimed to critically show the feasibility of using coal as a repository of CO₂, given its unique properties, especially the ability to perform CO₂ / CH₄ exchange sorption.

In order to meet the international protocols of climate changing up to 2050, technology of ECBM must advance progressively, since there is still immaturity for effective implementation of it.

Keywords: Climate changing, CO₂ sequestration, Enhanced Coal Bed Methane Recovery, CO₂/CH₄ exchange sorption

Introduction

The technology of CO₂ sequestration in deep coal seams have been research from decades [Fulton et al., 1980; Reznik et al., 1982], and it is still being developed.

In recent years, researchers at The Strata Mechanics Research Institute of the Polish Academy of Sciences have been carrying out studies connected with the issue concerning CO₂/CH₄ exchange sorption [Skoczylas and Kudasik, 2017; Kudasik et al., 2017; Dutka et al., 2013; Topolnicki et al., 2013; Dutka et al., 2012a; 2012b]. For the purposes of the studies many original measurement equipment were constructed which was used to carry out a series of experiments.

The detailed analysis of the literature on the subject showed that CO₂/CH₄ exchange sorption is well recognised. However, carry out exchange sorption in coal material under confining pressure (as in situ conditions) is a completely uninvestigated subject. Combining these two research directions is the subject of the present work that is being developed in The Strata Mechanics Research Institute of the PAS and financed from the resources of the National Science Centre in Poland, as part of the project entitled “CO₂/CH₄ exchange sorption in coal material under confining pressure”.

The purpose of this work was review about methodology of CO₂ storage in geological formation, especially coalbeds, in the context of the international climate change agreements.

The carbon cycle and global warming

Most of Earth's carbon – about 65,500 Gt is stored in rocks, however, it can also be found in the ocean, atmosphere, terrestrial and marine biosphere, as reservoirs. Through the geosphere, chemical reactions and tectonic activity, an average of 0.01-0.1 Gt of carbon move in the slow carbon cycle every year; and between 1-100 Gt of carbon move through the biosphere in the fast carbon cycle every year [Riebeek, 2011].

In recent decades, the CO₂ concentration in the atmosphere has increased by about 43%, from 280 ppm in the pre-industrial period, through 315 ppm in 1958 [Keeling, 1960] to over 400 ppm in 2015 [Dugugkenky and Tans, 2016]. The Earth that spent 10,000 years to warm up from the glacial to the interglacial phase (Holocene), since the last two decades has been heating almost 0.2°C per decade. This is a rate 50 times faster than the natural glacial-interglacial cycle [Nobre et al., 2012].

According to BP energy statistics, in the year 2017, there were 33 Gt of CO₂ emission worldwide. According to Nobre et al. [2012], the increase in carbon observed in the atmosphere is the organic origin CO₂, not the inorganic origin CO₂ released when a volcano erupts. Evaluation of isotopic carbon signatures lead to fossil sources and forest burning as responsible for the increase the concentration of this gas in the atmosphere, as a results of burning of coal, oil and natural gas by human activities.

Approaching climate change is a large-scale global challenge to reduce and avoid the release of enormous amounts of greenhouse gases (GHGs) over the course of this century. CO₂ is recognized as one of the principal greenhouse gases, which may be sequestered underground [Metz et al., 2005]. An assessment conducted by The Intergovernmental Panel of Climate Change [IPCC, 2014], concluded that the CO₂ emissions should be decreased by at least 50%, from the currently annual rate, to limit the escalation of the global average temperature to 2°C by 2050. This is a limit agreed by 125 nations that signed the Paris Agreement in 2015 – United Nations Framework Convention on Climate Change – UNFCCC – that would stabilize CO₂ concentration in the atmosphere “at a level that would prevent dangerous anthropogenic interference with the climate system”.

For IPCC assessment report, Smith et al. [2009], the calculation of climate change risk involves the process of integrated assessment, taking theoretical climate model-based projections of future climate change, using appropriate approaches to assess the likely environmental, societal, and economic impacts.

Carbon Capture and Storage in geological formations

Considering the global populations and standards of living for the four different future socio-economic scenarios by IPCC, this challenge implies a significant change in the way that energy is produced and consumed around the globe. Although the energy scenarios predict an increase in the share of the renewables, in absolute numbers the fossil fuels will continue to provide most of our energy needs in the foreseeable future. This is simply because is not likely the growth in renewable energy will be able to keep up with our increasing energy demand associated with a growing world population in such a short period [Smith et al., 2014]. Thus it is necessary to deal with this global challenge, more than ever, on different fronts. Therefore, new technologies are needed to be developed as one of the critical methods to mitigate the global warming issue [Norhasyima and Mahlia, 2018].

Even the most positive scenario, still points to the need for negative emissions [Shepherd, 2009]. Considering these tendencies, is growing the need of technologies to activity mitigate the effects of climate change. It involves land management practices, accelerated weathering, albedo modification, and also technologies to reduce carbon levels from the atmosphere, as through large-scale carbon dioxide removal – CDR [Boysen, 2017; Kreidenweis et al., 2016; Smith et al., 2014; Caldeira et al., 2013; Lackner et al., 2012; Vaughan and Lenton, 2011; Keith, 2009; Shepherd, 2009].

Thence, it is rising the development of carbon capture and storage – CCS – technologies has been oriented, as well as is included the reuse of CO₂ in the process, thus carbon capture utilization and storage – CCUS. Technology for CO₂ sequestration is still being developed, although some industrial-scale carbon sequestration projects are already operating around the world [Brennan et al., 2010; Sundquist et al., 2008].

CCUS is a methodology in which a relatively pure stream of CO₂ from industrial and energy-related sources is separated (captured), conditioned, compressed and transported to a storage location for long-term isolation from the atmosphere, but also produce valuable products, making this process more profitable and thus more viable. Due to the high cost and uncertainties in long-term geological storage, there is a growing inclination to include utilization, which re-use the CO₂, hence CCUS [Norhasyima and Mahlia, 2018].

In the context of flue gases, CO₂ is seen as a waste product. However, there are many applications, where CO₂ is utilized or considered as a valuable commodity [Smith et al., 2014]. IPCC [2014] stated that without CCUS implementation, the overall cost required to mitigate global climate change may increase up to 138% and there is a great challenge to achieve the 2°C targeted.

Given the enormous amounts of CO₂ that are emitting the geological sequestration emerges as a possible solution. Appropriate geological formations such as deep saline aquifers, depleted oil and gas fields, unmineable coal seams, and silicate formations can accommodate up to 11,000 Gt of CO₂ [Dooley et al., 2006]. For Dooley et al. [2006] this potential capacity should be more than enough to address global CO₂ storage needs for at least this century.

Many researches demonstrated that potential deep geologic CO₂ storage sites exist around the world. Although they are highly heterogeneous in quantity, quality and distribution, in many places, these candidates are near large groupings of power plants and other industrial facilities, which should lower the cost of deploying CCS systems [Dooley et al., 2006].

The concept of geological storage has been known for a long time. Similar deep geological formations have been used for oil and gas production and for fluid storage for over a century. But only recently the researchers related the potential of these formations and the climate change issue.

Dooley et al., [2006] explains the general conception of CO₂ storage reservoirs candidates. The reservoirs are normally at depths greater than 800 meters, sufficiently distant from water resources. CO₂ is injected as a supercritical fluid (dense as a liquid, but viscous like a gas), which allows it to flow easily through pipelines and into the geological formation. The layers of rocks that receive the CO₂ are more porous, facilitating the injection and storage of this. The surrounding rock layers are much more impermeable, working as reservoir seals, to keep CO₂ injected in place.

Many of the technologies needed to safely inject CO₂ into these deep geological formations already exist and are drawn from technologies, techniques and best industrial practices that are routinely used in the oil and gas industries. While CO₂ injection can be considered an established technology, large-scale deployment of CCS systems is potentially so great that it requires continuous development and demonstration in the field of more advanced techniques drilling and CO₂ injection [Dooley et al., 2006].

Coal as a CO₂ storage

For Mazumder and Wolf [2008] and Cui et al. [2003] coal, due to its properties, is a rock that under favourable geological conditions is the most promising CO₂ storage. Considering that the potential storage sites are any porous reservoir rocks with suitable impermeable cap rock, the coal seams that are not economically viable or too dangerous to mining can be considered to sequester CO₂ [Mukherjee and Misra, 2018]. Estimates [Metz et al., 2005] of CO₂ storage possible in global unmineable coal seams lie in the range of 3-200 Gt of CO₂, for Dooley et al [2006] it can be considered 140 Gt of CO₂.

As methane – CH₄ – is the main gas existent in the coal seams, it has been identified as an economical potential solution for the emerging world energy crisis. Because of the unique coal formation process compared to other reservoir rocks, the reservoir characteristics and gas storage and transfer mechanisms in coal reservoirs are quite different and therefore stand as important sources of natural gas [Sampath et al., 2017]. An additional advantage of this storage method is the recovery of methane (ECBM – enhanced coal bed methane recovery), thus it is characterized as an important method of carbon capture utilization and storage – CCUS. Norhasyima and Mahlia [2018] have methodically reviewed the patents for CO₂ utilization technologies for the application of CCUS, they have indicated the ECBM as one of the main technologies developed worldwide.

The CO₂ utilization potential should be of a scale proportionate with future CO₂ capture technology and requirements from large industrial sources and power generation. Potential CO₂ market demand and utilization method are presented in Table 1.

ECBM technology is a consequence of selective carbon sorption relative to CO₂. As was said, methane is found on the surfaces of the porous of coal. However, those surfaces have a chemical preference for CO₂ than CH₄. In case of low-pressure values, this relationship may stand at 8:1 [Krooss et al., 2002; Rodrigues and Lemos de Sousa, 2002]. When CO₂ is injected it induces the coal to release methane while adsorbing the injected CO₂ instead, it is called CO₂/CH₄ exchange sorption. The laboratory experiments are considered an effective approach in the characterization and quantification of the combined micro-processes that occur in ECBM and CO₂ sequestration [Norhasyima and Mahlia, 2018].

The analyses that phenomena in coal samples free of stress have been carried out since the 1980s. Fulton et al. [1980] presented the first laboratory tests of injecting CO₂ into coal samples to displace CH₄. Further analyses of the competitive adsorption/desorption process on coal were also presented by Reznik

Tab. 1. Potential CO₂ market demand. Comparison of ECBM with other technologies for carbon utilization [Norhasyima and Mahlia, 2018]

CO ₂ utilization method	Potential CO ₂ demand (Gt of CO ₂ / year)
Enhanced oil recovery (EOR) & Enhanced bed methane (ECBM)	0.3-3
Mineralization	>3
Fuel & Chemical including urea yield boosting	>3
Biofuel from algae	>3
Enhanced geothermal system (EGS)	0.05-0.3
Beverage carbonation	(about) 0.14
Food processing, packaging	(about) 0.15
Horticulture	0.01-0.05
Water treatment	0.01-0.05
Power generation – CO ₂ as working fluid	<0.01

et al. [1982], in higher pressure. The majority of laboratory analyses of CO₂/CH₄ exchange sorption have described the specific interactions between coal, CH₄ and CO₂, as well as identified some features of coal as a potential container of CO₂ and the source of CH₄ [Merkel et al., 2015; Zhang et al., 2015; Yu et al., 2014; Lafortune et al., 2014; Dutka et al., 2012; 2013; Baran et al., 2010; Majewska et al., 2009; Busch et al., 2003; 2006; Krooss et al., 2002; Clarkson and Bustin, 2000]. Thus, CO₂-driven enhanced coalbed methane recovery – ECBM – with simultaneous CO₂ storage is an emerging technology [Mazumder and Wolf, 2008; Dooley et al 2006, Cui et al., 2003].

Extracting unrecoverable oil reserves is made by injecting, for example, polymers and surfactant into the reservoirs to remove the trapped oil in the rocks. Although there are currently several techniques for the application of ECBM, its productivity has been greatly impaired due to existing poor practice. Due to the associated significant environmental risks, economic problems and reduced commercial viability the conventional methods of gas extraction are no longer acceptable. This has boosted the development of gas and coal production improvement techniques [Mukherjee and Misra, 2018; Sampath et al., 2017].

In this context, the use of CO₂ has become popular because it may potentially be then stored permanently in the same reservoir after production has completed, thus there would have been an environmental gain in reducing CO₂ emissions into the atmosphere [Mukherjee and Misra, 2018; Sampath et al., 2017]. Among the CO₂ based ECBM techniques, CO₂/CH₄ gas exchange and hydraulic fracturing using CO₂ are two popular techniques that give promising results [Sampath et al., 2017]. However, still uncertainties related to the underground behaviour of CO₂.

The challenge of ECBM

In ECBM, the injection of the CO₂ underground is reported to be a well-proven technology, achieved TRL-9, as petroleum industries have been injecting CO₂ in geological formations for many years. However, CO₂ utilization environmental and health impact is not yet well studied due to its immaturity of the technologies thus further studies are much needed. There are then some practical issues.

Firstly, ECBM is very much dependent on location-specific, whereby CO₂ sources and reservoir will determine its viability and economics. This means there isn't a universal viability [Mukherjee and Misra, 2018].

Secondly, according to Advanced Resources International [ARI, 2010], the CO₂ able to be used in ECBM is limited to only 60 million tons (0.06 Gt of CO₂), it is much less than the total emissions of CO₂, which implies that ECBM can only be a partial solution. Whereas ECBM should be considered only for mitigation purposes, would continue fulfilling its role.

The third point that could put the technology into questions is whether ECBM gives a negative balance of CO₂ in the atmosphere. The argument is that because of ECBM is produced more oil, and hence, further increase anthropogenic CO₂ emissions. The recovered CH₄, when burned will emit CO₂ in the atmosphere, thus this is exactly the opposite of what has been discussed to deal with the issues of climate change and CO₂ emission reduction. However it has been estimated that the burning of 1 ton CH₄ can yield 2.27 tons CO₂. The global survey indicates 3 Gt – 200 Gt of CO₂ can be sequestered, while the recoverable volume

of CH₄ is about 50 trillion cubic meter [Mukherjee and Misra, 2018]. Ottiger et al. [2006] suggested that if all recovered CH₄ is burned, there will be a net CO₂ storage, based on CO₂ balance and cycle.

The fourth potential issue would be related to the leakage. For Dooley et al. [2006], CO₂ leakage from deep geologic formations is not principally about human health and welfare today. In a properly designed and well-managed carbon capture utilization and storage facility – CCUS –, the chance of appreciable CO₂ leakage from the deep geologic storage formation is very small. The concern relates to slow, undetected leakage and how that might impact the climate for future generations. The scientific challenge is to ensure that CO₂ stays safe in these places for thousands of years. The development of monitoring, verification and evaluation technologies for all that CO₂ remains trapped in the soil is essential [Smith et al., 2014].

The principal task for the measurement, monitoring, and verification of stored CO₂ centers on how to demonstrate the long-term retention of stored CO₂ to regulators and the public. New and improved measurement and monitoring techniques and standards for their use need to be developed to provide proof of public and environmental safety and of each CCUS project's effectiveness in mitigating climate change [Dooley et al., 2006]. Ultimately, for Ramanov et al. [2009], the acceptance and implementation of any large-scale carbon sequestration methodology will be a social and political decision.

The expansion of this technology is needed at a fast pace to significantly contribute to the expectations of the century. This technology could play a significant role in society's efforts to stabilize atmospheric concentrations of greenhouse gases by the middle of the century. That means implanting thousands of carbon sequestration plants over the course of the century and as soon as possible. The implementation of this technology and effective implementation is not impossible, but definitely a challenge to reach the international protocol's targets.

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References

- Advanced Resources International (ARI), 2010: *U.S. Oil Production Potential from Accelerated Deployment of Carbon Capture and Storage*. 2010 Available at: <http://www.adv-res.com/pdf/v4ARI%20CCS-CO2-EOR%20whitepaper%20FINAL%204-2-10.pdf> (Accessed 11/09/2018).
- Baran P., Broś M., Nodzeński A., 2010: *Studies on CO₂ sorption on hard coal in the near-critical area with regard to the aspect of sequestration*. Archives of Mining Sciences, Vol. 55, Issue 1, 2010, Pages 59-68.
- Boysen L.R.W., Lucht D., Gerten V., Heck T., Lenton M., Schellnhuber H.J., 2017: *The limits to global-warming mitigation by terrestrial carbon removal*. Earth's Future, Vol. 5, 2017, Pages 463-474.
- BP, CO₂ emissions, 2017: <https://www.bp.com/en/global/corporate/energy-economics/statistical-review-of-world-energy/co2-emissions.html>. (Accessed 05/09/2018).
- Brennan S.T., Burruss R.C., Merrill M.D., Freeman P.A., Ruppert L.F., 2010: *A probabilistic assessment methodology for the evaluation of geologic carbon dioxide storage*. U.S. Geological Survey Open-File Report, 2010-1127, 2010.
- Busch A., Gensterblum Y., Krooss B.M., Siemons N., 2006: *Investigation of high-pressure selective adsorption/desorption CO₂ and CH₄ on coals: an experimental study*. International Journal of Coal Geology, Vol. 66, Issue 1-2, 2006, Pages 53-68.
- Busch A., Krooss B. M., Gensterblum Y., van Bergen F. Pagnier H.J.M., 2003: *High-pressure adsorption of methane, carbon dioxide and their mixtures on coals with a special focus on the preferential sorption behaviour*. Journal of Geochemical Exploration, Vol. 78-79, 2003, Pages 671-674.
- Caldeira K., Bala G., Cao L., 2013: *The science of geoengineering*. 2013 Annu. Rev. Earth Planet. Sci., 41, 231-256. <https://doi.org/10.1146/annurev-earth-042711-105548>.
- Clarkson C.R., Bustin R.M., 2000: *Binary gas adsorption/desorption isotherms: effect of moisture and coal composition upon carbon dioxide selectivity over methane*. International Journal of Coal Geology, Vol. 42, Issue 4, 2000, Pages 241-271.
- Cui X., Bustin R.M., Dipple G., 2003: *Selective transport of CO₂, CH₄ and N₂ in coals: insights from modeling of experimental gas adsorption data*. Fuel, Vol. 83, 2003, Pages 293-303.
- Dlugokencky E., Tans P., 2016: *Trends in atmospheric carbon dioxide*. National Oceanic & Atmospheric Administration, Earth System Research Laboratory (NOAA/ESRL), available at: <http://www.esrl.noaa.gov/gmd/ccgg/trends/global.html>, last access: 28 October 2016.

- Dooley J.J., Dahowski R.T., Davidson C.L., Wise M.A., Gupta N., Kim S.H., 2006: *Carbon Dioxide Capture and Geologic Storage: A Core Element of a Global Energy Technology Strategy to Address Climate Change*. Global Energy Technology Strategy Program, 2006.
- Dutka B., Kudasik M., Pokryszka Z., Skoczylas N., Topolnicki J., Wierzbicki M., 2013: *Balance of CO₂/CH₄ exchange sorption in a coal briquette*. Fuel Processing Technology, Vol. 106, 2013, Pages 95-101.
- Dutka B., Kudasik M., Skoczylas N., Wierzbicki M., 2012: *Laboratoryjne badania sorpcji wymiennej CO₂/CH₄ na brykiecie węglowym*. Prace Instytutu Mechaniki Górotworu PAN, Tom 14, nr 1-4, 2012, s. 15-24.
- Dutka B., Kudasik M., Topolnicki J., 2012: *Pore pressure changes accompanying exchange sorption of CO₂/CH₄ in a coal briquette*. Fuel Processing Technology, Vol. 100, 2012, Pages 30-34.
- Dutka B., Kudasik M., Topolnicki J., 2013: *Model numeryczny procesu sorpcji wymiennej CO₂/CH₄*. Prace Instytutu Mechaniki Górotworu PAN, Tom 15, nr 3-4, 2013, s. 3-14.
- Fulton P.F., Parente C.A., Rogers B.A., Shah N., Reznik A.A., 1980: *A laboratory investigation of enhanced recovery of methane from coal by carbon dioxide injection*. Proceedings of the SPE Unconventional Gas Recovery Symposium, May 1980; Pittsburgh, Pa, USA.
- IPCC, Climate change 2014: Synthesis Report. Contribution of Working Groups I, II and III to the Fifth Assessment Report of Intergovernmental Panel on Climate Change Geneva, Switzerland, 2014.
- Keeling C.D., 1960: *The Concentration and isotopic Abundance of Carbon Dioxide in the Atmosphere*. Tellus. 1960.
- Keith D., 2009: *Why Capture CO₂ From The Atmosphere*. Science, 325, Pp. 1654-1655.
- Kreidenweis U., Humpenöder F., Stevanović M., Bodirsky B.L., Kriegler E., Lotze-Campen H., Popp A., 2016: *Afforestation to mitigate climate change: Impacts on food prices under consideration of albedo effects*. 2016 Environ. Res. Lett., 11, 085001. <https://doi.org/10.1088/1748-9326/11/8/085001>.
- Krooss B.M., van Bergen F., Gensterblum Y., Siemons N., Pagnier H.J.M., David P., 2002: *High-pressure methane and carbon dioxide adsorption on dry and moisture-equilibrated Pennsylvanian coals*. International Journal of Coal Geology, Vol. 51, 2002, Pages 69-92.
- Kudasik M., Dutka B., Topolnicki J., 2012: *Investigating CO₂/CH₄ exchange sorption on coal*. Annual Report 2012, Polish Academy of Sciences, Pages 83-85.
- Kudasik M., Pajdak A., Skoczylas N., Wierzbicki M., 2017: *The studies of competitive sorption of CH₄-CO₂ under isothermal-isobaric conditions on coal* (Badania konkurencyjnej sorpcji CO₂ i CH₄ w warunkach izotermiczno-izobarycznych na węglu kamiennym). Prace Instytutu Mechaniki Górotworu PAN. 2017.
- Lackner K.S., Brennan S., Matter J.M., Park A.H.A., Wright A., Van Der Zwaan B., 2012: *The urgency of the development of CO₂ capture from ambient air*. 2012 Proc. Natl. Acad. Sci. U.S.A. 109, 13156–13162. doi:10.1073/pnas.1108765109.
- Lafortune S., Adeline F., Bentivegna G., Didier C., Farret R., Gombert P., Lagny C., Pokryszka Z., Canto Toimil C., 2014: *An experimental approach to adsorption of CO₂ + CH₄ gas mixtures onto coal (European RFCS CARBOLAB research project)*. Energy Procedia. 2014.
- Majewska Z., Ceglarska-Stefańska G., Majewski S., Ziętek J., 2009: *Binary gas sorption/desorption experiments on a bituminous coal: Simultaneous measurements on sorption kinetics, volumetric strain and acoustic emission*. International Journal of Coal Geology, Vol. 77, Issue 1-2, 2009, Pages 90-102.
- Mazumder S., Wolf K.H., 2008: *Differential swelling and permeability change of coal in response to CO₂ injection for ECBM*. International Journal of Coal Geology, Vol. 74, 2008, Pages 123-138.
- Merkel A., Gensterblum Y., Krooss B., Amann A., 2015: *Competitive sorption of CH₄, CO₂ and H₂O on natural coals of different rank*. International Journal of Coal Geology, 2015.
- Metz B., Davidson O., de Coninck H.C., Loos M., Meyer L.A., IPCC, 2005: *IPCC Special Report on Carbon Dioxide Capture and Storage. Working Group III of the Intergovernmental Panel on Climate Change*. Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA, 442 pp.
- Mukherjee M., Misra S., 2018: *A review of experimental research on Enhanced Coal Bed Methane (ECBM) recovery via CO₂ sequestration*. Earth-Science Reviews 2018.
- Nobre C.A., Reid J., Veiga A.P.S., 2012: *Scientific basis of climate change (Fundamentos científicos das mudanças climáticas)*. São José dos Campos, SP: Rede Clima/INPE, 2012. 44 p. ISBN: 978-85-17-00064-5.
- Norhasyima R.S., Mahlia T.M.I., 2018: *Advances in CO₂ utilization technology: A patent landscape review*. Journal of CO₂ Utilization (2018) 323-335.
- Reznik A., Singh P., Foley W., 1982: *Enhanced recovery of in situ methane by carbon dioxide injection: an experimental feasibility study*. Chemical and Petroleum Engineering Department, University of Pittsburgh, Pittsburgh, Pennsylvania, 1982.
- Riebeck H. Earth Observatory Nasa, 2011. <https://earthobservatory.nasa.gov/Features/CarbonCycle> (Accessed 11/09/2018).
- Rodrigues C.F., Lemos de Sousa M.J., 2002: *The measurement of coal porosity with different gases*. International Journal of Coal Geology, Vol. 48, 2002, Pages 245-251

- Sampath K.H.S.M., Perera M.S.A., Ranjith P.G., Matthai S.K., Rathnaweera T., Zhang G., Tao X., 2017: *CH₄eCO₂ gas exchange and supercritical CO₂ based hydraulic fracturing as CBM production-accelerating techniques: A review*. Journal of CO₂ Utilization. 2017.
- Shepherd J.G., 2009: *Geoengineering the climate: Science, governance and uncertainty*. 2009
- Skoczylas N., Kudasik M., 2017: *Sorpcja wymienna CO₂/CH₄ na węglu kamiennym poddanym obciążeniu okólnemu – analiza zagadnienia*. Prace Instytutu Mechaniki Górotworu PAN, Tom 19, nr 3, 2017, s. 47-54.
- Smit B.; Park A.H.A.; Gadikota G., 2014: *The grand challenges in carbon capture, utilization, and storage*. Front. Energy Res. 2014.
- Smith J.B., Schneider S.H., Oppenheimer M., Yohe G.W., Haref W., Mastrandrea M.D., Patwardhan A., Burton I., Corfee-Morlot J., Magadza C.H.D., Fussell H.-M., Pittcock A.B., Rahman A., Suarez A., Yperselen J.-P.v., 2009: *Assessing dangerous climate change through an update of the Intergovernmental Panel on Climate Change (IPCC) “reasons for concern”*. Proc Natl Acad Sci USA 106:4133-4137.
- Sundquist E., Burruss R., Faulkner S., Gleason R., Harden J., Kharaka Y., Tieszen L., Waldrop M., 2008: *Carbon sequestration to mitigate climate change: U.S. Geological Survey Fact Sheet 2008, 3097, 4 p.*
- Vaughan N.E., Lenton T.M., 2011: *A review of climate geoengineering proposals*. Climate Change, 109, 745-790.
- Yu H., Jing R., Wang P., Chen L., Yang Y., 2014: *Preferential adsorption behaviour of CH₄ and CO₂ on high-rank coal from Qinshui Basin, China*. International Journal of Mining Science and Technology, Vol. 24, Issue 4, 2014, Pages 491-497.
- Zhang J., Liu K., Clennell M.B., Dewhurst D.N., Pervukhina M., 2015: *Molecular simulation of CO₂-CH₄ competitive adsorption and induced coal swelling*. Fuel, 2015.

Rozwój technologii sekwestracji CO₂ w pokładach węgla – przegląd zagadnienia

Streszczenie

Biorąc pod uwagę globalny wyścig w redukcji emisji CO₂ w atmosferze, zalecona jest głównie ujemna emisja CO₂. W tym celu rozpoczęto na szeroką skalę badania nad sekwestracją CO₂ w pokładach węgla za pomocą technologii ECBM.

Praca jest przeglądem, który miał na celu pokazanie możliwości wykorzystania węgla jako składowiska CO₂, ze względu na jego wyjątkowe właściwości, w szczególności zdolność do procesu wymiany sorpcyjnej CO₂/CH₄.

Aby sprostać międzynarodowym standardom zmian klimatycznych do 2050 roku, technologia ECBM musi ciągle być rozwijana dla jej efektywnej implementacji.

Słowa kluczowe: zmiana klimatu, sekwestracja CO₂, technologia ECBM, sorpcja wymienna CO₂/CH₄